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Swerilactones C and D, anti-HBV New Lactones from a Traditional Chinese Herb: *Swertia mileensis*

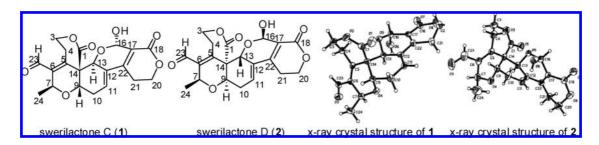
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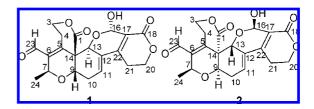
ABSTRACT



Swerilactones C (1) and D (2), two novel diastereomeric lactones with an unprecedented 6/6/6/6/6 pentacyclic ring system, were isolated from the traditional Chinese herb *Swertia mileensis*. Their structures and relative stereochemistry were elucidated on the basis of spectroscopic methods and further confirmed by X-ray single-crystal diffraction analysis. In vitro antihepatitis B virus (HBV) assay on the Hep G 2.2.15 cell line showed that both compounds 1 and 2 exhibited inhibitory activities against the secretion of HBsAg (IC₅₀ = 1.24 and 2.96 mM, respectively) and HBeAg (IC₅₀ = 0.77 and 1.47 mM, respectively).

As reported in our previous paper, *Swertia mileensis* (= *Swertia leducii*, generally known as "Qing-Ye-Dan" in Chinese) was discovered from the Yi and Hani regions and has been documented in *Chinese Pharmacopoeia* (1977–2005 editions) as a traditional Chinese medicine (TCM) to treat viral hepatitis. Primary anti-HBV assay in vitro showed the 90% and 50% EtOH extracts of the title plant possessed inhibitory activities on HBsAg and HBeAg. In a continuous search for active compounds, our further phytochemical investigation on this plant resulted in the isolation of another pair of novel diastereomeric lactones swerilactones C (1) and D (2), which were structurally different from the previously reported swerilactones A and B. This paper describes the

isolation and structural elucidation of compounds 1 and 2 by extensive spectroscopic and single-crystal X-ray crystallographic analyses, as well as their anti-HBV activities.



The previously isolated fraction A4 $(10.2 \text{ g})^1$ was chromatographed on a silica gel column $(150.0 \text{ g}, 4.0 \times 25.0 \text{ g})$

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Table 1. ¹H (500 MHz) and ¹³C (125 MHz) NMR Data of Compounds 1 and 2 in Pyridine- d_5 (δ in ppm, J in Hz)

position	$1\delta_{\mathrm{H}}$	$1\delta_{\mathrm{C}}$	$oldsymbol{2}$ $\delta_{ ext{H}}$	$2~\delta_{ m C}$
1		167.2, s		167.2, s
3	4.68 (1H, m)	67.8, t	4.75 (1H, m)	67.4, t
	4.32 (1H, m)		4.28 (1H, m)	
4	3.69 (1H, m)	25.0, t	3.61 (1H, m)	25.4, t
	3.63 (1H, m)		3.55 (1H, m)	
5		154.0, s		153.4, s
6		136.0, s		136.3, s
7	4.93 (1H, q, 6.6)	69.0, d	4.85 (1H, q, 6.3)	71.6, d
9	4.50 (1H, d, 4.3)	66.4, d	3.99 (1H, d, 4.0)	72.1, d
10	3.44 (1H, dd, 20.2, 4.1)	33.4, t	3.38 (1H, dd, 20.1, 4.0)	33.2, t
	2.53 (1H, dd, 20.2, 4.1)		2.58 (1H, dd, 20.1, 4.0)	
11	6.33 (1H, bs)	130.8, d	6.31 (1H, bs)	130.9, d
12		127.9, s		127.8, s
13	6.54 (1H, s)	71.5, d	6.62 (1H, s)	71.8, d
14		$50.5, \mathrm{s}$		51.1, s
16	6.46 (1H, d, 3.8)	88.3, d	6.46 (1H, s)	88.2, d
17		123.0, s		122.9, s
18		164.0, s		164.0, s
20	4.24 (1H, m)	65.6, t	4.22 (1H, m)	65.6, t
	4.14 (1H, m)		4.07 (1H, m)	
21	2.49 (1H, m)	23.2, t	2.48 (1H, m)	23.0, t
	2.32 (1H, m)		2.28 (1H, m)	
22		144.8, s		144.7, s
23	10.26 (1H, s)	189.5, d	$10.26(1\mathrm{H,s})$	190.5, d
24	1.50 (3H, d, 6.6)	19.2, q	1.43 (3H, d, 6.4)	21.0, q

cm) eluting with CHCl₃/Me₂CO (90:10 \rightarrow 30:70) to supply subfractions 1–5. The subfraction 2 (1.5 g) was chromatographed on silica gel column (30.0 g, 2.0 \times 23.0 cm) with an eluent of CH₃Cl/MeOH (95:5), and further purified with RP-18 column chromatography (120.0 g, 2.8 \times 28.0 cm, eluted with MeOH/H₂O, 80:20) to afford swertilactone C (1) (30 mg) and swertilactone D (2) (10 mg).

Swerilactone C (1)² was isolated as colorless cubic crystals (Me_2CO) , $[\alpha]^{28.2}$ _D +14.66 (c 0.20, CHCl₃/MeOH v/v = 3:1). Its molecular formula was determined to be C₂₀H₂₀O₈ on the basis of (-) HRESIMS (calcd for $[M + C1]^-$ m/z 423.0846, found 423.0843) with 11 degrees of unsaturation. The IR spectrum of compound 1 exhibited OH (3455 cm⁻¹), C=O (1717 cm⁻¹), and double bond (1644 cm⁻¹), respectively. The ¹H and ¹³C NMR spectra of compound 1 (Table 1) showed 20 carbon resonances due to eight quaternary carbons, six tertiary carbons, five methylenes, and one methyl group. Among them, an aldehyde group ($\delta_{\rm C}$ 189.5; $\delta_{\rm H}$ 10.26, s), two lactone carbonyl carbons ($\delta_{\rm C}$ 167.2 and 164.0), six olefinic carbons ($\delta_{\rm C}$ 154.0, 144.8, 136.0, 130.8, 127.9, 123.0), and one dioxygenated carbon ($\delta_{\rm C}$ 88.3; $\delta_{\rm H}$ 6.46, d, J=3.8Hz) were deduced. Accordingly, a pentacyclic ring structure was required for compound 1 to fulfill the unsaturation requirement.

The partial structure **1a** (Figure 1) was established by ${}^{1}H^{-1}H$ COSY (H-24/H-7; H-3/H-4) and HMBC (H-23/C-6, C-7; H-7/C-6, C-5 and H-3/C-5, C-1). Similarly, the fragment **1b** [${}^{1}H^{-1}H$ COSY (H-9/H-10/ H-11), HMBC (H-10/C-12; H-13/C-11)] and **1c** [${}^{1}H^{-1}H$ COSY (H-20/H-21), HMBC (H-16/C-17, C-18, C-22); H-20/C-18, C-22; H-21/C-17] were also determined, respectively. In addition, the linkages of C(7)—O—C(9) and C(13)—O—C(16) were deduced by the detection of the HMBC correlations from H-7 to C-9 and from H-16 to C-13. The other correlations in the HMBC spectrum of compound **1** were revealed as follows: H-13 ($\delta_{\rm H}$ 6.54, s) with C-1 and C-14; H-9 ($\delta_{\rm H}$ 4.50, d, J = 4.3 Hz) with C-1, C-5, C-11, and C-13, which led to the

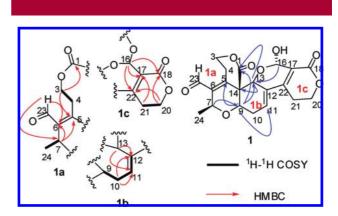


Figure 1. Fragment structures and key COSY and HMBC correlations of compound 1.

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⁽²⁾ Compound 1: mp 255–256 °C; $[\alpha]^{28.2}_{\rm D}$ +14.66 (c 0.20, CHCly/MeOH v/v = 3:1); UV (MeOH) $\lambda_{\rm max}$ (log ε) 377 (3.14), 256 (4.32) nm; IR (KBr) $\lambda'_{\rm max}$: 3455, 1717, 1644, 1434, 1394, 1355, 1176, 1089, 1062, 1031, 983, 808 cm⁻¹; ¹H and ¹³C NMR data, see Table 1; EIMS m/z 388 ([M]⁺, 1), 370 (41), 253 (23), 195 (100), 177 (72), 165 (47), 149 (59); ESIMS (–) m/z 423 ([M + Cl]⁻); HRESIMS (–) m/z 423.0843 [M + Cl]⁻ ($C_{20}H_{20}O_8Cl$ calcd 423.0846).

connection of C-1, C-5, C-9, and C-13 with C-14. Thus, the planar structure of compound **1** was determined as shown in Figure 1.

In the ROESY spectrum (Figure 2), the correlations of H-24/H-9, H-11/H-21, and H-23/H-4 were observed. Nevertheless, the above-mentioned evidence could not provide sufficient information to confirm the stereochemistry of compound **1**. Thus, a single X-ray crystallographic analysis was conducted, which not only verified the deduced planar structure of compound **1**, but also determined its relative stereochemistry as shown in Figure 3.³ Based on the IUPAC nomenclature rule the relative stereocenters of C-7, 9, 13, 14, 16 were deduced as S^* , S^* , S^* , R^* , S^* , respectively.

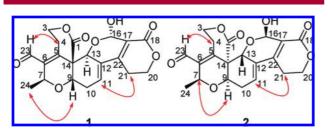


Figure 2. Selected ROESY correlations of 1 and 2.

Swerilactone D (2)⁴ had the same molecular formula as that of swerilactone C (1) deduced from its (–) HRESIMS [calcd for [M – H]⁻ m/z 387.1079, found 387.1079]. Its IR spectrum displayed absorptions of OH (3334 cm⁻¹), C=O (1727, 1701 cm⁻¹), and double bond (1637 cm⁻¹). Detailed comparison of their NMR data indicated that they were a pair of isomers with the main difference of the downfield shift of C-9 from δ_C 66.4 (d) in compound 1 to δ_C 72.1 (d) in compound 2 and the upfield shift of H-9 from δ_H 4.50 (1H, d, J = 4.3 Hz) in compound 1 to δ_H 3.99 (1H, d, J = 4.0 Hz) in compound 2. With the HMBC spectrum, the planar structure of compound 2 was determined to be identical with that of compound 1. In its ROESY spectrum (Figure 3), the correlations H-9/H-7, H-11/H-21, H-23/H-4 were observed, instead of the corelation H-9/H-24 observed

in compound 1, and the detected correlation H-9/H-7 indicated the different stereochemistry between their structures. With the above deduction, it is still insufficient to conform the stereochemistry of compound 2. Therefore, a single X-ray diffraction study was performed and its relative stereochemistry was determined to be 7S*,9R*,13R*,14S*,16R* (Figure 3).⁵

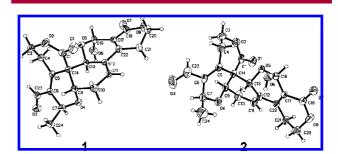


Figure 3. X-ray crystal structures of compounds 1 and 2.

Swerilactones C (1) and D (2) were another pair of novel diastereomeric lactones isolated from S. mileensis. Compared to swerilactones A and B, swerilactones C (1) and D (2) also possessed a 6/6/6/6 pentacyclic ring system skeleton, but there was little similarity between their chemical structures. Interestingly, swerilactones A-D contained a δ -lactone fragment similar to those of the aglycon part of secoiridoidal glycosides, a family of compounds widely presented in Gentianaceae. 6-8 S. mileensis belonging to the Swertia genus (Gentianaceae) was reported as being rich in secoiridoids (iridoids). 9,10 Thus, we presumed that swerilactones A-D would be biogenetically generated from secoiridoids. However, this specific biogenetic pathway is still unclear and needs further investigation. Compounds 1 and 2 were tested for their anti-HBV activities in vitro on the HBV-transfected Hep G 2.2.15 cell line as reported previously. 11,12 Both swerilacones C and D exhibited inhibitory activities against the secretion of HBsAg ($IC_{50} = 1.24$ and

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⁽³⁾ Crystallographic data of compound 1: $C_{20}H_{20}O_8$, MW = 388.36; monoclinic, space group $p2_1$; a=6.547(2) Å, b=10.892(3) Å, c=12.418(4) Å, $\alpha=90.00$, $\beta=103.542(4)$, $\gamma=90.00$, V=860.9(5) ų, Z=2, d=1.498 g/cm³, crystal dimensions $0.16\times0.12\times0.08$ nm³ were used for measurement on a SHELXL-97 with a graphite monochromater, Mo K α radiation. The total number of reflections measured was 3852, of which 1635 were observed, $I>2\sigma(I)$. Final indices: $R_1=0.0681$, $wR_2=0.1487$. The crystal structure of compound 1 was solved by direct method SHLXS-97 (Sheldrick, 1990) and expanded using the difference Fourier technique, refined by the program SHLXL-97 (Sheldrick, 1997), and the full-matrix least-squares calculations. Crystallographic data for the structure of compound 1 have been deposited with the Cambridge Crystallographic Data Centre (deposition no. CCDC 732831). Copies of these data can be obtained free of charge via www.ccdc.cam.ac.uk/conts/retrieving.htm. (or from the Cambridge Crystallographic Data Centre, 12, Union Road, Cambridge CB21EZ, U.K.; fax (+44) 1223-336-033; or desposit@ccdc.cam.ac.uk).

⁽⁴⁾ Compound 2: mp 240–241 °C; $[\alpha]^{28.9}_{D}$ –12.90 (*c* 0.22, CHCl₃/MeOH v/v = 1:1); UV (MeOH) λ_{max} (log ε) 262 (4.23) nm; IR (KBr) λ'_{max} 3334, 1727, 1701, 1660, 1637, 1449, 1407, 1316, 1177, 1090, 1067, 1030, 982, 809 cm⁻¹; ¹H and ¹³C NMR data, see Table 1; ESIMS (–) m/z 387 ([M – H]⁻); (–) HRESIMS m/z 387.1079 [M – H]⁻ ($C_{20}H_{19}O_{8}$ calcd 387.1079).

⁽⁵⁾ Crystallographic data of compound **2**: $C_{20}H_{20}O_8$ (H₂O), MW = 406.38; triclinic, space group p1; a=8.6844(14) Å, b=14.094(2) Å, c=17.000(3) Å, $\alpha = 73.293(2)$, $\beta = 87.688(2)$, $\gamma = 72.958(2)$, V = 1903.2(5) Å^3 , Z = 4, $d = 1.418 \text{ g/cm}^3$, crystal dimensions $0.20 \times 0.14 \times 0.08 \text{ nm}^3$ were used for measurement on a SHELXL-97 with a graphite monochromater, Mo $K\alpha$ radiation. The total number of reflections measured was 8035, of which 3235 were observed, $I > 2\sigma(I)$. Final indices: $R_1 = 0.0409$, $wR_2 = 0.1000$. The crystal structure of compound 2 was solved by direct method SHLXS-97 (Sheldrick, 1990) and expanded using the difference Fourier technique, refined by the program SHLXL-97 (Sheldrick, 1997), and the full-matrix least-squares calculations. Crystallographic data for the structure of compound 2 have been deposited with the Cambridge Crystallographic Data Centre (deposition no. CCDC 737797). Copies of these data can be obtained free of charge via www.ccdc.cam.ac.uk/conts/ retrieving.htm (or from the Cambridge Crystallographic Data Centre, 12, Union Road, Cambridge CB21EZ, U.K.; fax (+44) 1223-336-033; or desposit@ccdc.cam.ac.uk).

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2.96 mM, respectively) and HBeAg (IC₅₀ = 0.77 and 1.47 mM, respectively. 13

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Supporting Information Available: 1D and 2D NMR, $[\alpha]_D$, UV, IR, MS spectra, and X-ray crystallographic data (CIF) of swerilactones C (1) and D (2) and experimental procedures. This material is available free of charge via Internet at http://pubs.acs.org.

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^{(13) 3}TC (lamivudine, an antiviral agent) was used as the positive control in our anti-HBV screening. 3TC showed inhibitory activity against HBsAg secretion (IC $_{50} = 20.1$ mM, SI = 1.3) and against HBeAg secretion (IC $_{50} = 30.5$ mM, SI < 1).